

# Designing a Hybrid Science Lab Course: General Chemistry I

Aisling M. O'Connor

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# It's a Changing World!



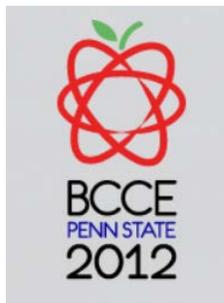
# Early Interest in Distance Education!



# How I Got Started!



- 2006: International marketing on-line



- 2012: Designing, development and teaching of on-line chemistry courses



- 2014: Distance learning & hybrid teaching mini-workshop

# General Chemistry I

- Introductory course for science majors
- Taken by many EXSS students & those interested in careers in the health sciences
- 2 lectures (1 hr. 15 min.) & 1 lab (2 hr. 45 min.) per week
- Challenging course for students with weak math skills
- Offered every semester and in Summer I
- Summer I students are generally strong & motivated

# General Chemistry I Hybrid Lab Course

- Lecture portion of course all on-line
- All lectures recording using Tech Smith Relay
- Tablet used to annotate PowerPoint slides and solve problems
- Class meets 3 hours per week / 9 hours per week in Summer I (15 sessions)
  - Orientation & safety training
  - 10 experiments
  - 4 – 5 problem sheets
  - 2 review sessions
  - 3 exams
- Sapling on-line homework
- Taught as a Summer I course & GCE spring night course

# Schedule for Typical Week (Summer I)

Sessions	Lecture Recordings	Face-to-Face Meetings	Reading	Homework
<p><b><u>Week 3:</u></b> June 12 - 18</p> <p><b>Class Meetings:</b> Tues., June 13 Thurs., June 15 Fri., June 16</p>	<p>13. Types of Chemical Reactions Part III</p> <p>14. Solutions</p> <p>15. Quantitative analysis</p> <p>16. Light Theory</p> <p>17. Quantum Mechanics &amp; Numbers</p> <p>18. Electron Configuration</p>	<p><b><u>Tuesday, June 13:</u></b> Mid-term exam review session Problem Worksheet 3</p> <p><b><u>Thursday, June 15:</u></b> <i>Mid Term Exam, 9:30 – 11:30 am</i></p> <p><b><u>Friday, June 16:</u></b> <i>Experiment 6: Analysis of Vinegar</i></p>	<p>Chapter 4 (4.9)</p> <p>Chapter 4 (4.4)</p> <p>Chapter 7</p> <p>Chapter 8 (8.1 – 8.4)</p>	<p>Sapling On-line Homework 5, due June 14 @ 11:30 pm</p> <p>Sapling On-line Homework 6, due June 18 @ 11:30 pm</p>

# Blackboard Site

The screenshot displays a Blackboard course site for "Week 3 (June 12 - 18)". The browser address bar shows the URL: <https://blackboard.fitchburgstate.edu/webapps/blackboard/cont>. The user is identified as "Aisling O'Connor".

The course navigation includes "My Fitchburg State", "Courses", and "OneCard". The current course path is "Course Documents > Week 3 (June 12 - 18)".

The main content area is titled "Week 3 (June 12 - 18)" and features a navigation bar with "Build Content", "Assessments", "Tools", and "Partner Content".

The content list includes the following items:

- 13. Types of Chemical Reactions Part III** (Enabled: Statistics Tracking)
- 13 Types of Chemical Reactions Part III.pdf** (Enabled: Statistics Tracking)
- 14. Solutions** (Enabled: Statistics Tracking)
- 14 Solutions.pdf** (Enabled: Statistics Tracking)
- 15. Quantitative Analysis** (Enabled: Statistics Tracking)
- 15 Quantitative Analysis.pdf** (Enabled: Statistics Tracking)
- 16. Light Theory** (Enabled: Statistics Tracking)
- 16 Light Theory.pdf** (Enabled: Statistics Tracking)

The page is viewed at 75% zoom.

# Example Lecture

- Lecture 14: Solutions
- <http://www.screencast.com/t/cFlyfLfCDzXq>



## Preparing Solutions By Dilution Example

A stock solution of NaCl is 6.00 M. How much of this stock solution is needed to prepare 1.00-L of physiological saline solution (0.154 M)?

$$M_c V_c = M_d V_d$$

$$V_c = \frac{M_d V_d}{M_c} = \frac{(0.154 M)(1.00 L)}{6.00 M}$$

$$V_c = 0.0257 L$$

$$0.0257 L \frac{1000 \text{ mL}}{1 L} = 25.7 \text{ mL}$$

# Due Dates / Calendar

Due Date / Reminder Calendar, May / June 2017

Sun	Mon	Tue	Wed	Thu	Fri	Sat
28	29	30 <b>Class</b> First class meeting	31 H/W 1 due @ 11:30 pm	1 <b>Class</b> Pre- lab Q. Exp. 1 due	2 <b>Class</b> Pre- lab Q. Exp. 2 & Lab Report Exp. 1 due	3
4 H/W 2 due @ 11:30 pm	5	6 <b>Class</b> Pre- lab Q. Exp. 3 & Lab Report Exp. 2 due	7 H/W 3 due @ 11:30 pm	8 <b>Class</b> Pre- lab Q. Exp. 4 & Lab Report Exp. 3 due <i>Early Term Exam</i>	9 <b>Class</b> Pre- lab Q. Exp. 5 & Lab Report Exp. 4 due	10
11 H/W 4 due @ 11:30 pm	12	13 <b>Class</b> Lab Report Exp. 5 due	14 H/W 5 due @ 11:30 pm	15 <b>Class</b> <i>Mid-Term Exam</i>	16 <b>Class</b> Pre- lab Q. Exp. 6	17
18 H/W 6 due @ 11:30 pm	19	20 <b>Class</b> Pre- lab Q. Exp. 7 & Lab Report Exp. 6 due	21 H/W 7 due @ 11:30 pm	22 <b>Class</b> Pre- lab Q. Exp. 8 & Lab Report Exp. 7 due	23 <b>Class</b> Pre- lab Q. Exp. 9 & Lab Report Exp. 8 due	24 <i>Take Home Exam Due @ 11:30 pm</i>
25 H/W 8 due @ 11:30 pm	26	27 <b>Class</b> Pre- lab Q. Exp. 10 & Lab Report Exp. 9 due	28 H/W 9 due @ 11:30 pm	29 <b>Class</b> Lab Report Exp. 10 due	30 <b>Class</b> Last class meeting <i>Final Exam</i>	1

# Pros / Cons

- Students spend less time on campus
  - Increased flexibility
  - Still have face-to-face access to professor
  - Interact with peers in lab
  - More in-class time for problem solving and review
  - On-campus exams (2 – 3 hours)
  - Recorded lectures can be posted for traditional Gen. Chem. I course
- Students not taking the time to view / study lecture material
  - Cannot ask questions during lecture
  - Weaker students may struggle more
  - Spring GCE courses not as successful as Summer I courses

# Lessons Learned /Plans for the Future

- Constant reminders regarding due dates and keeping up with lecture material are necessary
- Due date / reminder calendar has worked very well
- Lecture recording are too long!
- Need a method to ensure students are viewing / studying lectures
- Re-record lectures; split up from 24 to approx. 50
- Short Blackboard quiz to be completed after each lecture
- Hybrid Forensic Chemistry class (non-lab)?
- Help others develop hybrid science lab courses

# Acknowledgements

- Dr. Mike Leamy
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